

XII - II - Revision - ChemistryAnswer KeyPart - I15 x 1 = 15

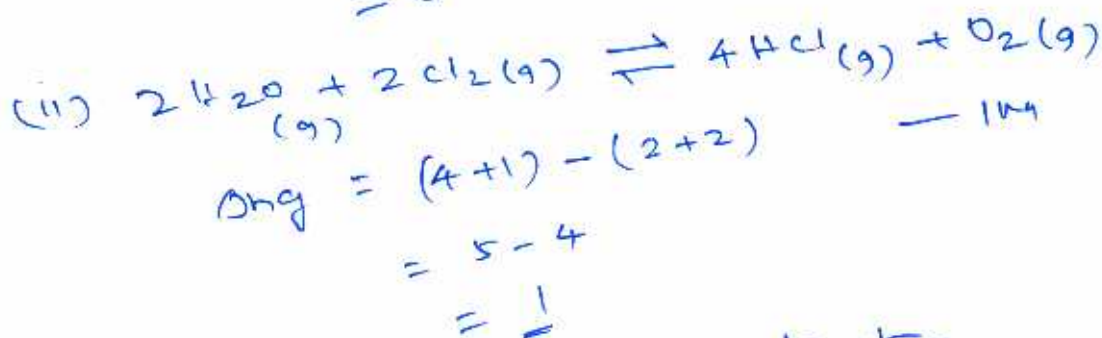
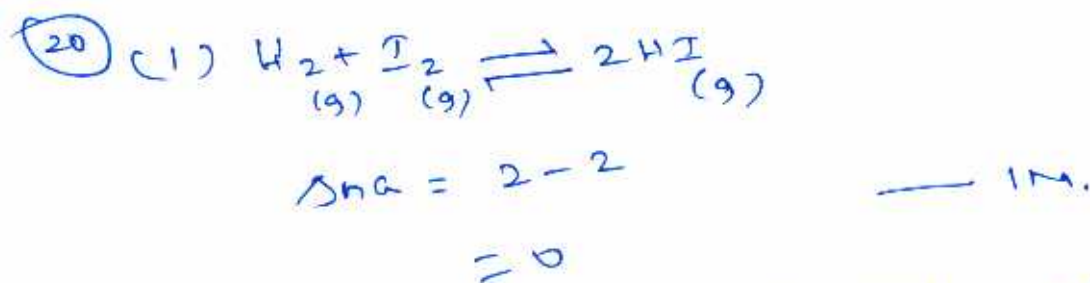
- ① - 4E (b)
- ② Ar F₆ (d)
- ③ Ag Br (a)
- ④ Transition metals (a)
- ⑤ Fe²⁺ (b)
- ⑥ all the above (d)
- ⑦ $q = 0$ (b)
- ⑧ (c) $k = Ae^{-E_a/RT}$
- ⑨ (b) scattering of light
- ⑩ (a) CH₃COOH
- ⑪ (b) 2-pentanone
- ⑫ (b) acetaldehyde
- ⑬ (a) Calcium oxalate
- ⑭ (c) C₆H₅N₂Cl
- ⑮ (d) water

Part - II

- ⑯ Correct Statement of - 2M
Heisenberg uncertainty Principle.
- ⑰ (i) Bond energies of many elements are not known - 1M
(ii) electron affinities of many elements are not known - 1M

(18) Correct Statement of Splitting of Silver
 - 2M.

(19) Ultra Cold Substances Conductor
 Current at zero resistance (or)
 without resistance
 - 2M.



(21) Correct Statement of Activation Energy
 - 2M

(or)

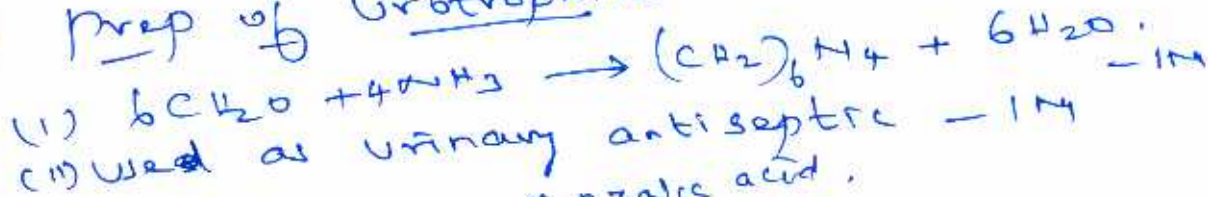
$$A.E = \bar{A}.E - \text{Energy of Colliding molecules.}$$

 - 1M

(22) Two Conditions of Optically activity.

- (i) Asymmetric Carbon - 1M
- (ii) Chirality - Non super Imposable mirror Image - 1M

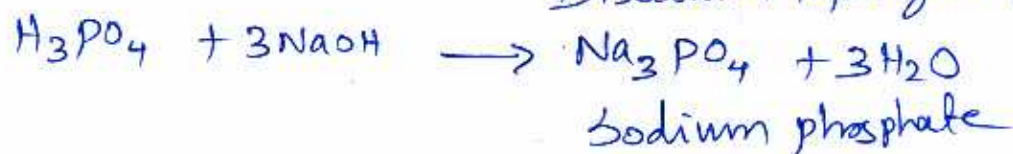
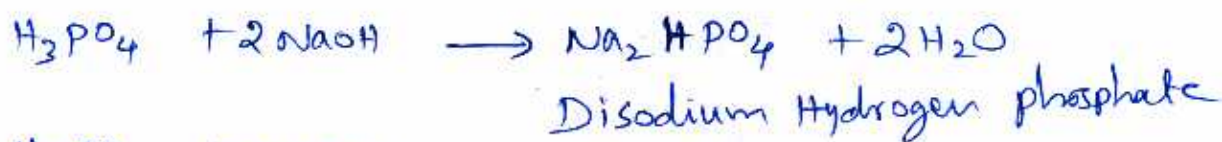
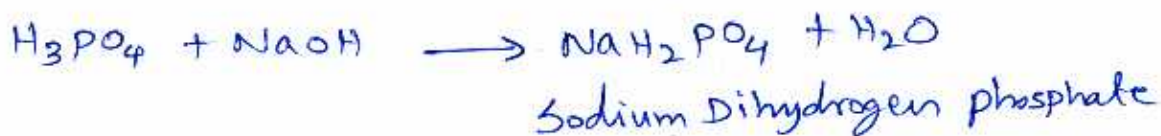
(23) Prep of Urotropine



(24) Any two uses of oxalic acid.
 - 2M.

PART - III

25. H_3PO_4 is triprotic - 3 equations (3 marks)



26. Chrome plating correct explanation with anode and cathode.

Cathode - Article to be plated with chromium ($\frac{1}{2}$ m)

Anode - Lead plate ($\frac{1}{2}$ m)

Electrolyte - Chromic acid and Sulphuric acid ($\frac{1}{2}$ m)

Explanation - $1\frac{1}{2}$ mark

27. Lanthanide Contraction:

Correct Definition - 2 mark

Causes - 1 mark.

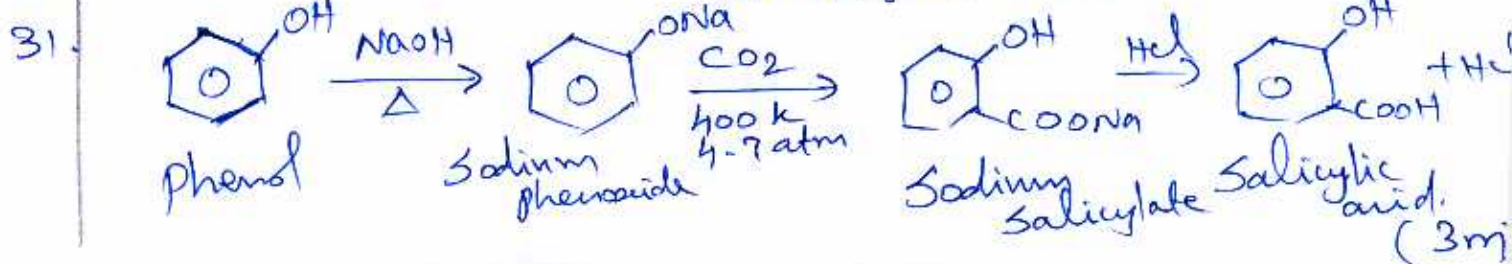
28. Entropy Definition - 2 mark

Unit of entropy - 1 mark

29. Tyndall effect correct explanation - 3 mark.

30. Common ion effect Definition - 2 mark

example - 1 mark.



32. Statement - 1 mark
 equation - 2 mark.

33. Antacids Definition - 2 mark
 example - 1 mark

PART-IV

34(a)

NaF crystal - $\text{Na}^+ = 2, 8$
 $\text{F}^- = 2, 8$ } Ne type configuration
 - 1 mark

$$\gamma_{\text{C}^+} + \gamma_{\text{A}^-} = d_{\text{C}^+\text{A}^-} \quad \text{--- 1 mark.}$$

Expansion

$$\gamma_{\text{C}^+} \propto \frac{1}{z^*(\text{C}^+)} \quad \text{--- 1 mark}$$

$$\gamma_{\text{A}^-} \propto \frac{1}{z^*(\text{A}^-)} \quad \text{--- 1 mark.}$$

$$\frac{\gamma_{\text{C}^+}}{\gamma_{\text{A}^-}} = \frac{z^*(\text{A}^-)}{z^*(\text{C}^+)} \quad \text{--- 1 mark.}$$

(b) Extraction ^(OR) of Zinc:

- Chief ore - $\frac{1}{2}$ mark.
 Concentration - $\frac{1}{2}$ mark
 Roasting - 1 mark.
 Reduction - 1 mark
 Purification - 2 mark.

35(a) Extraction of Lanthanides from monazite sand:

Flow chart - 3m

Separation - 2m

(02)
 (b) Various Statement of second law of thermodynamics.

Kelvin - Planck - 1 mark

Clausius - 1 mark

Entropy Statement - 1 mark

Efficiency of machine with explanation } 2 marks

36 (a) Derive k_p and k_c for PCl_5 Dissociation.

k_c Derivation - 2 1/2 marks

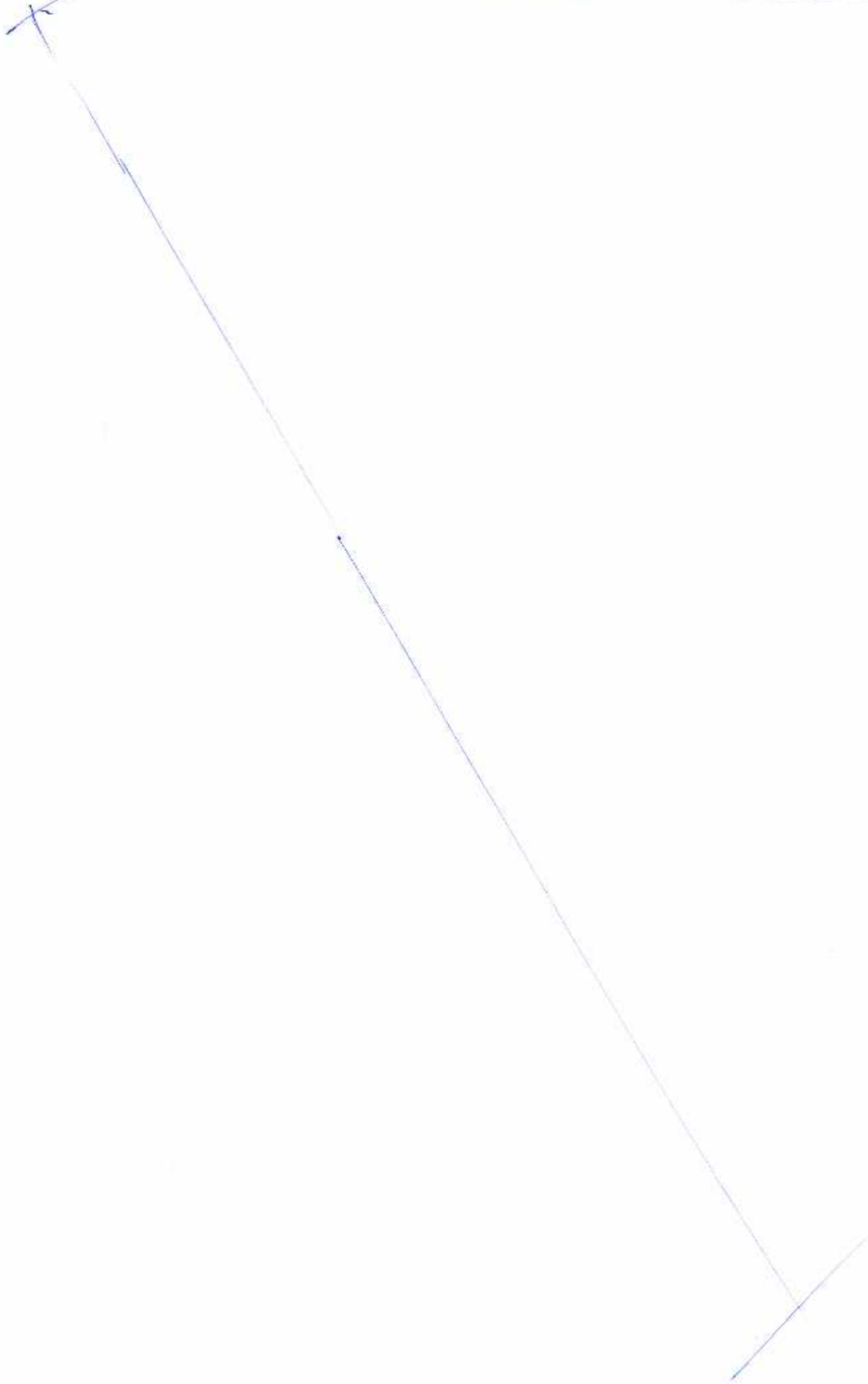
k_p Derivation - 2 1/2 marks

(02)

(b) Consecutive reactions - 2 marks

Parallel reaction - 1 1/2 marks

Opposing reaction - 1 1/2 marks



37. a) Derivation of Nernst equation.

$$A + B \rightleftharpoons C + D \quad - \frac{1}{2} \text{ mark}$$

$$-\Delta G^\circ = -\Delta G^\circ - RT \ln J \quad - \frac{1}{2} \text{ mark}$$

$$-\Delta G^\circ = -\Delta G^\circ - RT \ln \frac{a_C \times a_D}{a_A \times a_B} \quad - 1 \text{ mark}$$

$$E = E^\circ - \frac{RT}{nF} \ln \frac{a_C \times a_D}{a_A \times a_B} \quad - 1 \text{ mark}$$

$$E = E^\circ - \frac{RT}{nF} \ln \frac{[C][D]}{[A][B]} \quad - 1 \text{ mark}$$

$$E = E^\circ - 2.303 \frac{RT}{nF} \log K \quad - 1 \text{ mark}$$

(b) Preparation of anisole. (OR)
 - 3 preparations - 5 marks

38. (a) (i) Trans-esterification

Statement - 1 mark
 Equation - 1 1/2 marks

(ii) Sandmeyer reaction:

Statement - 1 mark.
 Equation - 1 1/2 marks

(b)(i) Zwitter ion :

Explanation - $1\frac{1}{2}$ mark

Formula - 1 mark

(ii) Analgesics :

Explanation - $1\frac{1}{2}$ mark

Example - 1 mark.

Handling Teachers

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30/11/14